

Supplementary Information for

Petrogenesis of Néma 001, an alkali-rich meteorite from the

acapulcoite-lodranite parent body

R. Tartèse, J. Gattacceca, G. Avice, P. Beck, B. Devouard, V. Debaille, L. Fawcett,
G. Hublet, K. H. Joy, C. Sonzogni, and J. Villeneuve

Supplementary Figures 1 to 7.

Supplementary Tables (available at <https://doi.org/10.48420/29126567.v3>)

Table S1. Bulk ICP-MS

Table S2. EPMA analyses

Table S3. LA-ICP-MS protocol - Traces

Table S4. LA-ICP-MS silicates - all

Table S5. LA-ICP-MS protocol - Dating

Table S6. Apatite U-Pb dating

Table S7. LA-ICP-MS apatite - all

Table S8. Al-Mg dating

Table S9. Ne isotopes

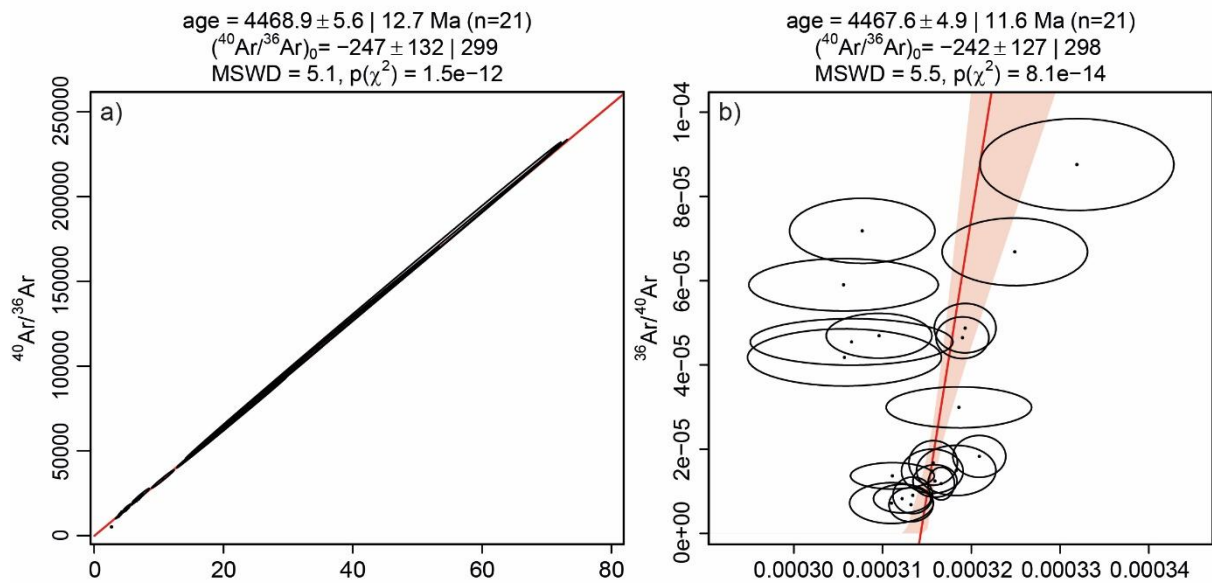
Table S10. Ar isotopes

Table S11. Mineral chemistry

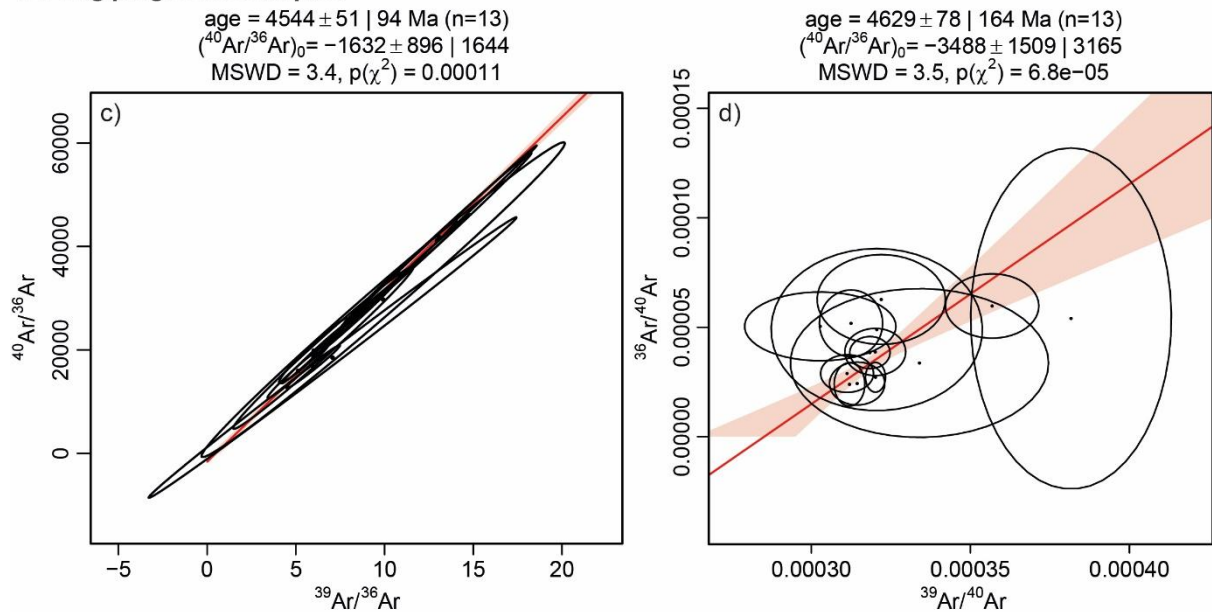
Table S12. MELTS modelling

Supplementary Figures

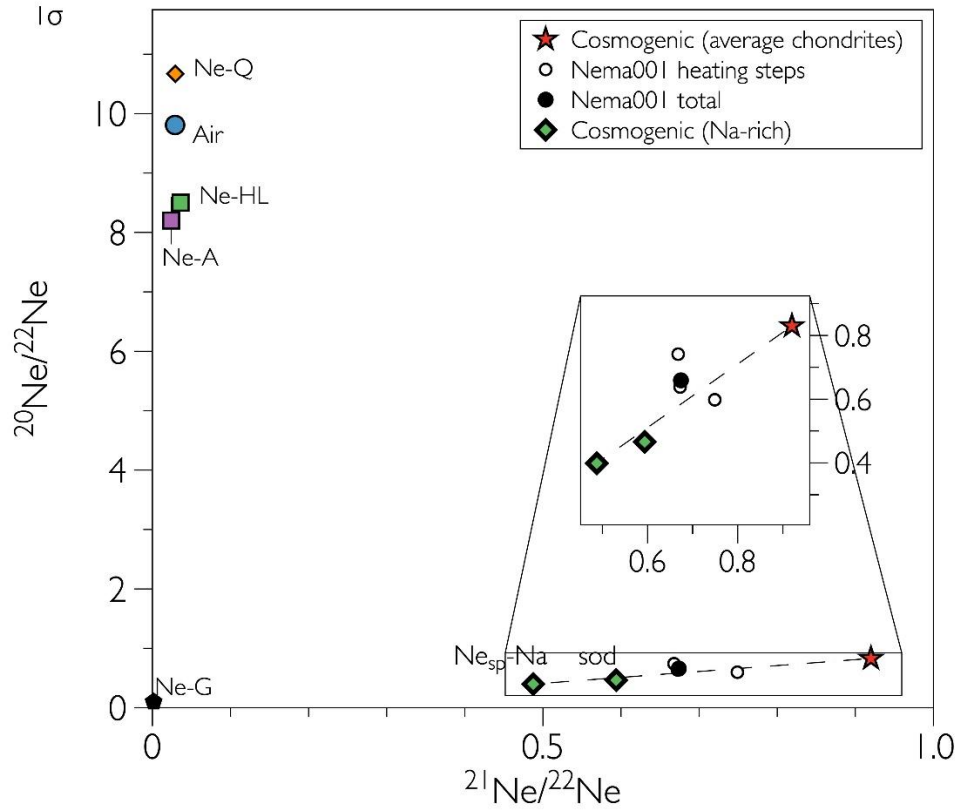
1.4 mg plagioclase aliquot



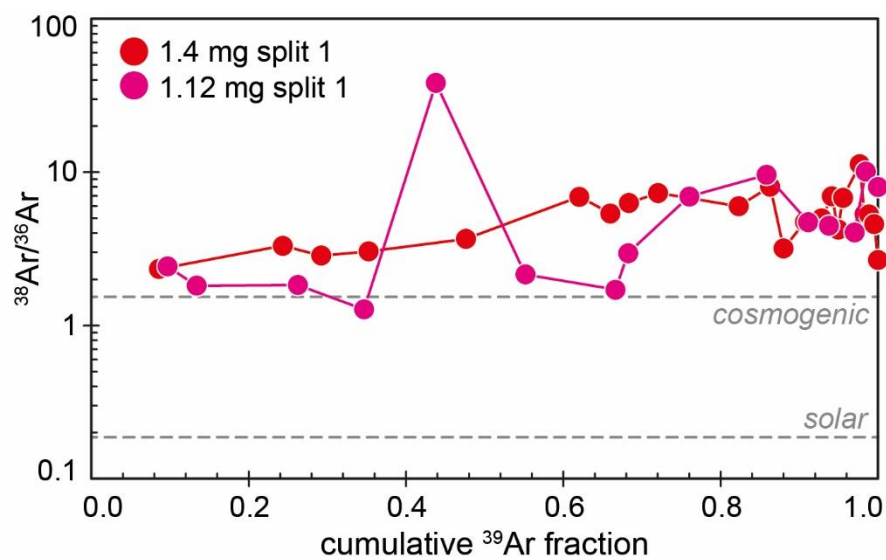
1.12 mg plagioclase aliquot



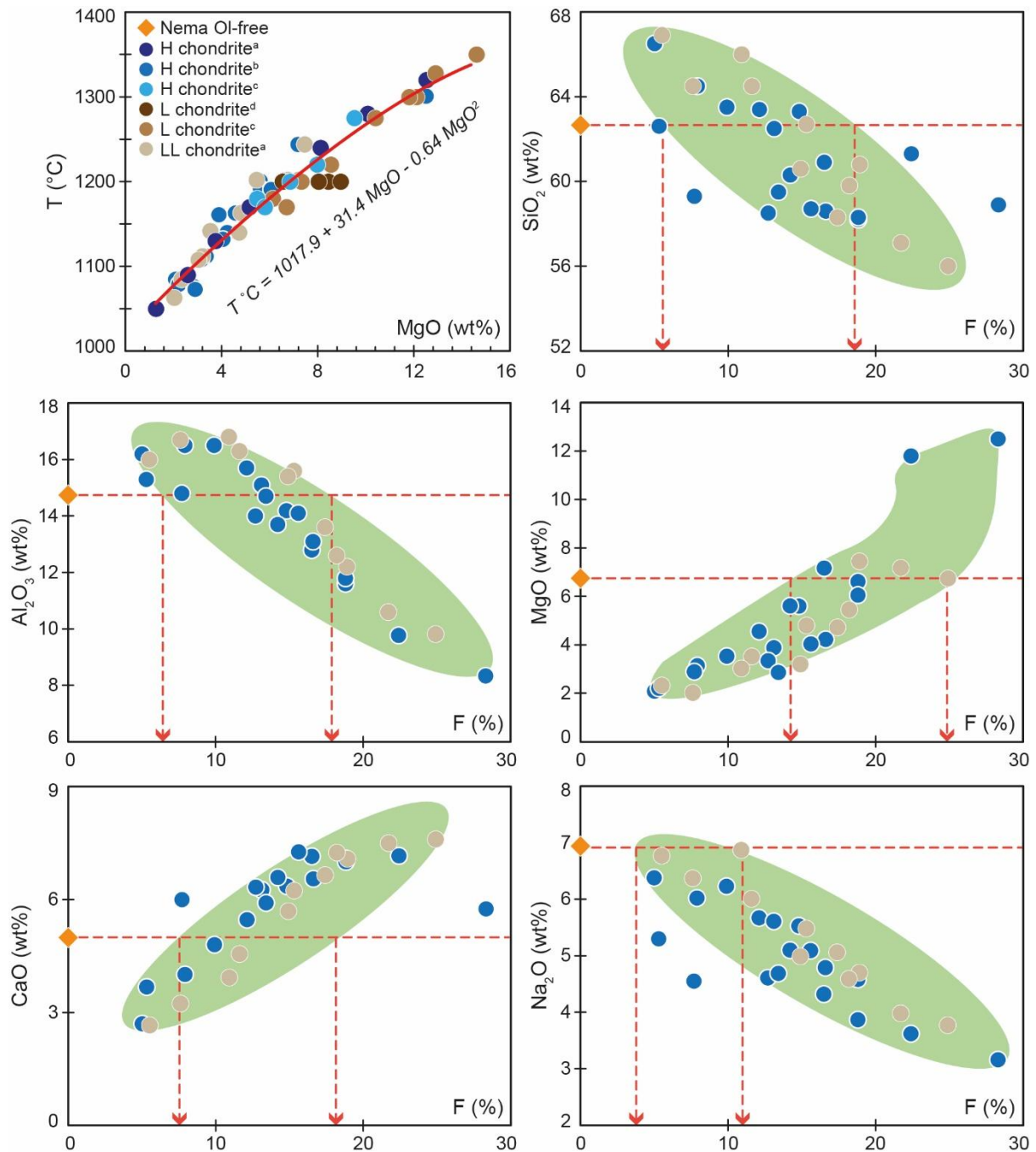
Supp. Figure 1: Normal (a, c) and inverse (b, d) Ar isochron plots for the two plagioclase splits.



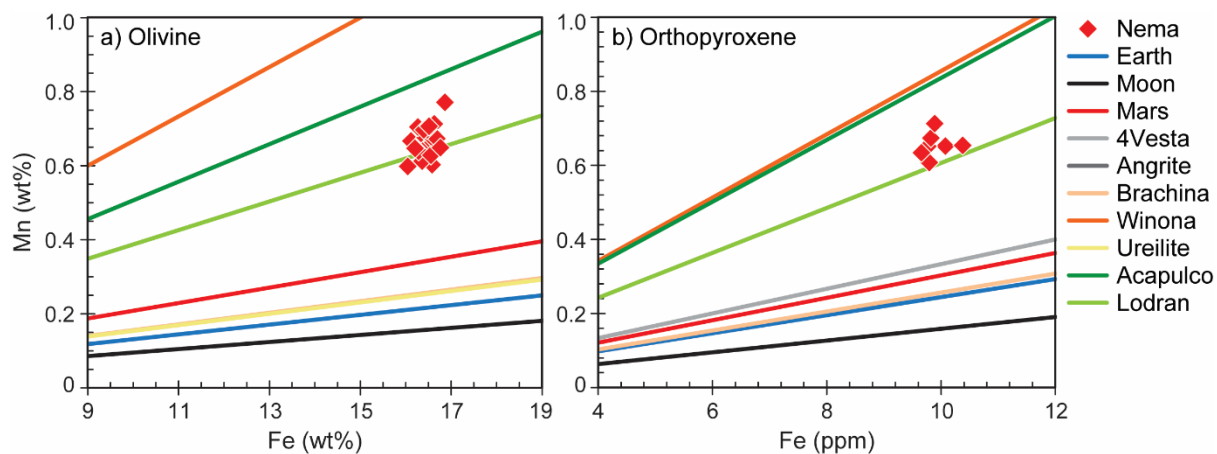
Supp. Figure 2: Three Ne isotope plot displaying the isotopic composition of neon released from a bulk aliquot of Néma 001 during laser heating steps (empty circles) and the isotopic composition of total neon measured at IPGP (filled circle). All heating steps plot between cosmogenic neon produced from spallation of pure Na (NeSp-Na) or sodalite (sod) (see Pravdivsteva et al., 2020 and references therein) and average cosmogenic neon measured in chondrites (Wieler, 2002). The isotope compositions for neon from phase Q (Ne-Q; Busemann et al., 2000), the Earth's atmosphere (Air; Ozima & Podosek, 2001), and other primordial components found in meteorites (Ne-HL, Ne-A, and Ne-G; Ott, 2014) are also plotted. Uncertainties are smaller than the size of the symbols.



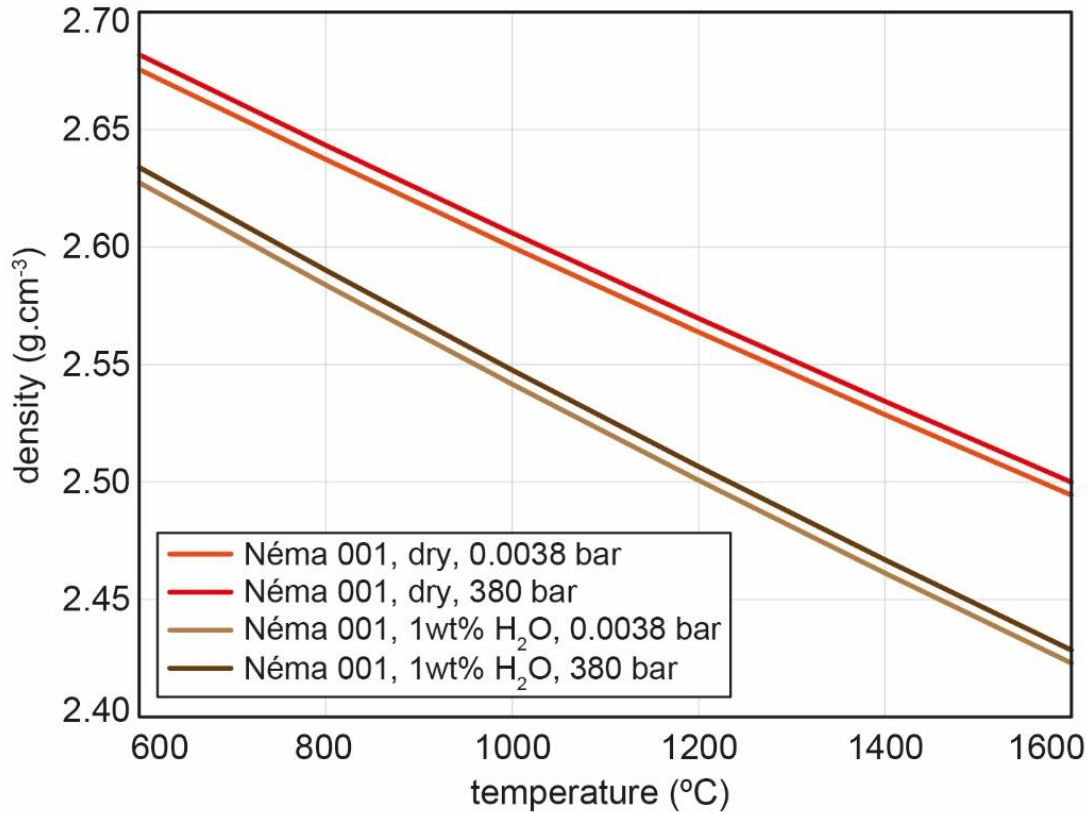
Supp. Figure 3: $^{38}\text{Ar}/^{36}\text{Ar}$ vs. cumulative ^{39}Ar fraction plot displaying the step-heating results for the two plagioclase splits of Néma 001. The cosmogenic and solar $^{38}\text{Ar}/^{36}\text{Ar}$ ratios are from Hennessy and Turner (1980).



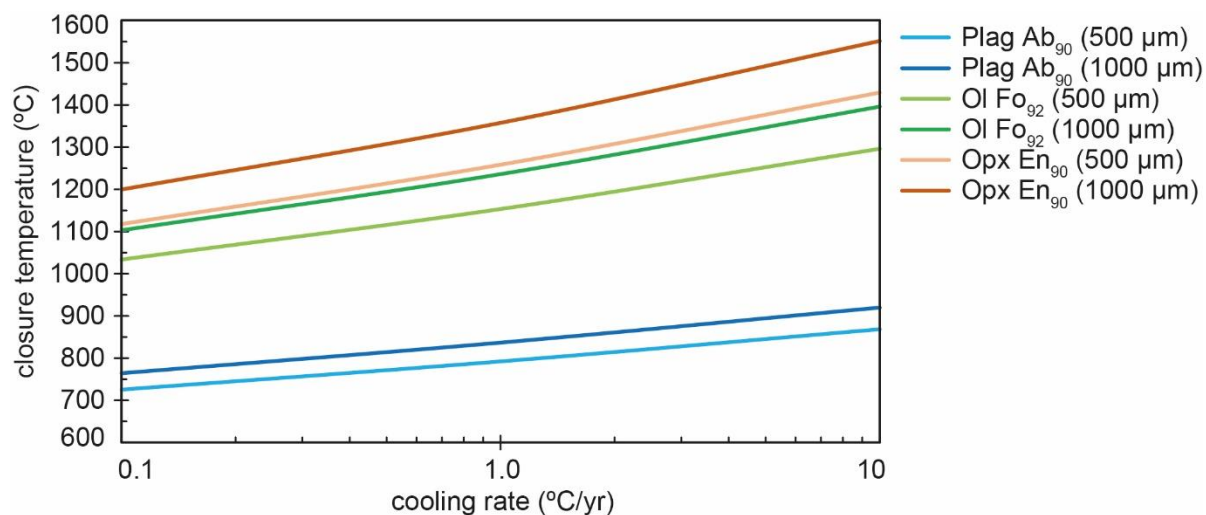
Supp. Figure 4: Temperature vs. bulk MgO abundance (top left) and bulk major oxide abundances vs. degree of partial melting (F) for silicate liquids produced by melting ordinary chondrite precursors at 1050-1350 °C and $f\text{O}_2$ between IW-2.4 and IW+0.8 (Jurewicz et al., 1995 (c); Feldstein et al., 2001 (d); Usui et al., 2015 (b); Collinet and Grove, 2020 (a)).



Supp. Figure 5: Mn (wt%) vs. Fe (wt%) abundance in olivine (a) and pyroxene (b) for Néma 001 compared to a range of planetary materials. References for Fe and Mn abundances are Papike et al. (2003) for Earth, Moon, Mars, and 4Vesta; Prinz et al. (1977) for the angrite Angra Dos Reis; Nehru et al. (1983) for Brachina; Benedix et al. (1998) for Winona; Berkley et al. (1980) for the ureilite Novo Urei; Keil and McCoy (2018) for Acapulco and Lodran.



Supp. Figure 6: Density, calculated using the DensityX programme (Iacovino and Till, 2019), for a melt with the major element composition of Néma 001, with either 0 or 1 wt% H₂O, for temperatures between 600 and 1600 °C, and lithostatic pressures of 0.0038 and 380 bars (corresponding to depths of 1 m and 100 km, respectively, on a putative parent body with a lithostatic column density of 3.8 g.cm⁻³ and a gravitational acceleration g of 0.1 m.s⁻²; e.g., Lucas et al., 2022).



75

76 Supp. Figure 7: Closure temperature calculated for Mg diffusion in Ab₉₀ plagioclase,
 77 Fo₉₂ olivine, and En₉₀ orthopyroxene for cooling rates between 0.1-10 °C/yr that
 78 corresponds to estimates for the first fast cooling phase down to ~750 °C for
 79 acapulcoites-lodranites (Lucas et al., 2022). Calculations carried out using the
 80 equation of Dodson (1973) for a cylinder, and Mg diffusion parameters of Van Orman et
 81 al. (2014) for plagioclase, Chakraborty et al. (1994) for olivine, and Schwandt et al.
 82 (1998) for orthopyroxene.

References

- Benedix, G. K., McCoy, T. J., Keil, K., Bogard, D. D. and Garrison, D. H. (1998) A petrologic and isotopic study of winonaites: Evidence for early partial melting, brecciation, and metamorphism. *Geochim. Cosmochim. Acta* 62, 2535-2553. [https://doi.org/10.1016/S0016-7037\(98\)00166-5](https://doi.org/10.1016/S0016-7037(98)00166-5).
- Berkley, J. L., Taylor, G. J., Keil, K., Harlow, G. E. and Prinz, M. (1980) The nature and origin of ureilites. *Geochim. Cosmochim. Acta* 44, 1579-1597. [https://doi.org/10.1016/0016-7037\(80\)90119-2](https://doi.org/10.1016/0016-7037(80)90119-2).
- Busemann, H., Baur, H. and Wieler, R. (2000) Primordial noble gases in “phase Q” in carbonaceous and ordinary chondrites studied by closed-system stepped etching. *Meteorit. Planet. Sci.* 35, 949-973. <https://doi.org/10.1111/j.1945-5100.2000.tb01485.x>.
- Chakraborty, S., Farver, J. R., Yund, R. A. and Rubie, D. C. (1994) Mg tracer diffusion in synthetic forsterite and San Carlos olivine as a function of P, T and fO₂. *Phys. Chem. Minerals* 21, 489-500. <https://doi.org/10.1007/BF00203923>.
- Collinet, M. and Grove, T. L. (2020) Widespread production of silica- and alkali-rich melts at the onset of planetesimal melting. *Geochim. Cosmochim. Acta* 277, 334-357. <https://doi.org/10.1016/j.gca.2020.03.005>.
- Dodson, M. H. (1973) Closure temperature in cooling geochronological and petrological systems. *Contrib. Mineral. Petrol.* 40, 259-264. <https://doi.org/10.1007/BF00373790>.
- Feldstein, S. N., Jones, R. H. and Papike, J. J. (2001) Disequilibrium partial melting experiments on the Leedey L6 chondrite: Textural controls on melting processes. *Meteorit. Planet. Sci.* 36, 1421-1441. <https://doi.org/10.1111/j.1945-5100.2001.tb01836.x>.
- Hennessy, J. and Turner, G. (1980) ⁴⁰Ar-³⁹Ar ages and irradiation history of Luna 24 basalts. *Phil. Trans. Roy. Soc. A* 297, 27-39. <https://doi.org/10.1098/rsta.1980.0202>.
- Iacovino, K. and Till, C. B. (2019) DensityX: A program for calculating the densities of magmatic liquids up to 1,627 °C and 30 kbar. *Volcanica* 2, 1-10. <https://doi.org/10.30909/vol.02.01.0110>.
- Jurewicz, A. J. G., Mittlefehldt, D. W. and Jones, J. H. (1995) Experimental partial melting of the St. Severin (LL) and Lost City (H) chondrites. *Geochim. Cosmochim. Acta* 59, 391-408. [https://doi.org/10.1016/0016-7037\(94\)00328-J](https://doi.org/10.1016/0016-7037(94)00328-J).
- Keil, K. and McCoy, T. J. (2018) Acapulcoite-lodranite meteorites: Ultramafic asteroidal partial melt residues. *Chem. Erde* 78, 153-203. <https://doi.org/10.1016/j.chemer.2017.04.004>.
- Lucas, M. P., Dygert, N., Ren, J., Hesse, M. A., Miller, N. R. and McSween, H. Y. (2022) Thermochemical evolution of the acapulcoite-lodranite parent body: Evidence for

fragmentation-disrupted partial differentiation. *Meteorit. Planet. Sci.* 57, 2248-2275.
<https://doi.org/10.1111/maps.13930>.

Nehru, C. E., Prinz, M., Delaney, J. S., Dreibus, G., Palme, H., Spettel, B. and Wänke, H. (1983) Brachina: a new type of meteorite, not a chassignite. *J. Geophys. Res.* 88, B237-B244. <https://doi.org/10.1029/JB088iS01p0B237>.

Ozima, M. and Podosek, F. A. (2001) *Noble Gas Geochemistry*, 300 pp., Cambridge University Press, Cambridge, UK. <https://doi.org/10.1017/CBO9780511545986>.

Ott, U. (2014) Planetary and pre-solar noble gases in meteorites. *Geochemistry* 74, 519-544. <https://doi.org/10.1016/j.chemer.2014.01.003>.

Papike, J. J., Karner, J. M. and Shearer, C.K. (2003) Determination of planetary basalt parentage: A simple technique using the electron microprobe. *Am. Mineral.* 88, 469-472. <https://doi.org/10.2138/am-2003-2-323>.

Pravdivtseva, O., Tissot, F. L. H., Dauphas, N. and Amari, S. (2020) Evidence of presolar SiC in the Allende Curious Marie calcium–aluminium-rich inclusion. *Nature Astro.* 4, 617-624. <https://doi.org/10.1038/s41550-019-1000-z>.

Prinz, M., Keil, K., Hlava, P. F., Berkley, J. L., Gomes, C. B. and Curvello, W. S. (1977) Studies of Brazilian Meteorites, III. Origin and history of the Angra dos Reis achondrite. *Earth Planet. Sci. Lett.* 35, 317-330. [https://doi.org/10.1016/0012-821X\(77\)90134-0](https://doi.org/10.1016/0012-821X(77)90134-0).

Schwandt, C. S., Cygan, R. T. and Westrich, H. R. (1998) Magnesium self-diffusion in orthoenstatite. *Contrib. Mineral. Petrol.* 130, 390-396.
<https://doi.org/10.1007/s004100050373>.

Usui, T., Jones, J. H. and Mittlefehldt, D. W. (2015) A partial melting study of an ordinary (H) chondrite composition with application to the unique achondrite Graves Nunataks 06128 and 06129. *Meteorit. Planet. Sci.* 50, 759-781.
<https://doi.org/10.1111/maps.12392>.

Van Orman, J. A., Cherniak, D. J. and Kita, N. T. (2014) Magnesium diffusion in plagioclase: dependence on composition, and implications for thermal resetting of the ^{26}Al – ^{26}Mg early solar system chronometer. *Earth Planet. Sci. Lett.* 385, 79-88.
<https://doi.org/10.1016/j.epsl.2013.10.026>.

Wieler, R. (2002) Cosmic-Ray-Produced Noble Gases in Meteorites. *Rev. Mineral. Geochem.* 47, 125-170. <https://doi.org/10.2138/rmg.2002.47.5>.